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Solid-State Molecular Dynamics of $(Ph₃Si)Si(SiMe₃)₃ Studied by Variable$ Temperature ¹³C and ²⁹Si MAS NMR **Spectroscopy**

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Summary. One- and two-dimensional ^{13}C and ^{29}Si solid-state NMR experiments at variable temperature demonstrate that crystalline $(Ph_3Si)Si(SiMe_3)$ ₃ lacks all molecular symmetry and that internal $2\pi/3$ reorientation of all three crystallographically inequivalent SiMe₃ groups in the molecule occurs. Quantitative analysis of one-dimensional MAS NMR spectra and of twodimensional NMR exchange experiments by spectral lineshape simulations yields exchange rate constants for internal SiMe_3 reorientation as a function of temperature. The activation energies determined from these kinetic data are $E_a = 21 \pm 4$, 22 ± 4 , and $30 \pm 5 \text{ kJ} \cdot \text{mol}^{-1}$ for internal reorientation of the three inequivalent SiMe_3 groups in solid (Ph₃Si)Si(SiMe₃)₃.

Keywords. Molecular solid-state dynamics; Solid-state NMR; 13C and 29Si MAS NMR; Exchange rate constants; Spectral lineshape simulations.

Untersuchung der molekularen Dynamik von $(Ph_3Si)Si(SiMe_3)_3$ im Festkörper mittels ¹³Cund 29Si-MAS-NMR-Spektroskopie bei variabler Temperatur

Zusammenfassung. Ein- und zweidimensionale 13 C- und 29 Si-NMR-Experimente am Festkörper bei verschiedenen Temperaturen zeigen, daß kristallines $(Ph_3Si)Si(SiMe_3)$ keine molekulare Symmetrie besitzt und daß interne $2\pi/3$ -Reorientierung der drei kristallographisch unterschiedlichen SiMe3-Gruppen auftritt. Quantitative Analyse von eindimensionalen MAS-NMR-Spektren und zweidimensionalen Austauschexperimenten mittels Linienformsimulation ergibt die Geschwindigkeitskonstanten des obengenannten Austauschs als Funktion der Temperatur. Daraus erhält man für die interne Reorientierung der drei nichtäquivalentem SiMe₃-Gruppen in festem (Ph₃Si)Si(SiMe₃)₃ die folgenden Aktivierungsenergien: 21 ± 4 , 22 ± 4 und 30 ± 5 kJ·mol⁻¹.

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Introduction

 $Si(SiMe₃)₄$ and $C(SiMe₃)₄$ are highly disordered crystalline phases at ambient temperatures, similar to the disordered crystalline phases which exist for several nonpolar, spherically shaped organic molecules such as e.g. adamantane [1]. Both compounds undergo structural phase transitions to less disordered low-temperature phases (Si(SiMe₃)₄ at $T < 238$ K [2], C(SiMe₃)₄ at $T < 213$ K [3]); the molecular dynamic properties of these two low-temperature phases have been investigated by variable temperature 13 C and 29 Si MAS NMR. The molecular point group symmetry in the low-temperature phases of $Si(SiMe₃)₄$ and $C(SiMe₃)₄$, as inferred from MAS NMR experiments, is C_3 [4-6] and is confirmed by X-ray diffraction analysis [7]. Despite identical molecular symmetry in these two low-temperature phases, the two compounds differ strongly with regard to the dynamic properties of their lowtemperature phases. Quantitative analysis of one- and two-dimensional variable temperature 13 C and 29 Si MAS NMR experiments permits to distinguish and precisely determine exchange rate constants for internal SiMe_3 reorientation and for whole-molecule reorientation around preferred molecular axes as a function of temperature. The dynamic solid-state properties of the low-temperature phases of $Si(SiMe₃)₄$ and $C(SiMe₃)₄$ may be briefly summarized as follows. In the lowtemperature phase of $Si(SiMe₃)₄$ we observe internal $SiMe₃$ reorientation and additional whole-molecule reorientation; the exchange rate constants for internal SiMe₃ reorientation always considerably exceed those for whole-molecule reorientation. The situation is different for solid $C(SiMe₃)₄$: over the entire temperature range of existence of the low-temperature phase, only whole-molecule reorientation is observed, whereas exchange rate constants for internal SiMe3 reorientation are negligibly small [6]. We have interpreted this difference in the solid-state molecular dynamic properties of $Si(SiMe₃)₄$ and $C(SiMe₃)₄$ in relation to higher intramolecular steric crowding in $C(SiMe₃)₄$. This interpretation is qualitatively supported by empirical force-field calculations based on the gas-phase structures of $Si(SiMe₃)₄$ [8] and $C(SiMe₃)₄$ [9].

Here we report on the solid-state dynamic properties of a compound chemically closely related to $Si(SiMe₃)₄$. In $(Ph₃Si)Si(SiMe₃)₃$ (1) the molecular symmetry is lowered by chemical substitution as compared to the fully SiMe_3 substituted parent molecule. We have investigated the dynamic solid-state properties of 1 by variable temperature one- and two-dimensional 13 C and 29 Si CP/MAS NMR experiments. The experimental NMR results on 1 are compared to the dynamic properties of the low-temperature phase of $Si(SiMe₃)₄$.

Results and Discussion

Apart from internal reorientation of the methyl groups themselves (which will not be considered in the following) and possible π -flips of the phenyl groups in the $SiPh₃$ moiety, *a priori* there are two modes of molecular dynamics to be considered for solid $(Ph₃Si)Si(SiMe₃)₃(1)$:

i) internal reorientation of the individual R_3 Si groups ($R = Ph$, Me) around the R_3 Si-Si_{central} bond directions, leading to mutual exchange of the groups R within a $SiR₃$ unit (referred to as internal reorientation), and

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ii) reorientation of the $Si(SiMe₃)₃$ moiety around the Ph₃Si-Si_{central} bond direction axis, leading to mutual exchange of the three $SiMe₃$ groups as a whole (referred to as $Si(SiMe₃)₃$ reorientation).

These two reorientational modes, if present, may occur simultaneously. Our experimental solid-state NMR tools to investigate and possibly distinguish these two reorientational modes consist of one- and two-dimensional variable temperature ${}^{13}C$ and ${}^{29}Si$ NMR experiments in conjunction with spectral lineshape simulations.

Many solid organosilicon compounds undergo structural phase transitions upon cooling or heating. Differential scanning calorimetry (DSC) of 1 gives no indication for the occurrence of such structural phase transitions in solid 1 in the temperature range of $T = 180-290$ K. Accordingly, observation of temperature dependent spectral lineshapes in 13C and 29Si MAS NMR experiments on 1 cannot be ascribed to the effects of structural phase transitions in this temperature range.

Under MAS conditions, that is at MAS rates sufficiently high to average out all ²⁹Si chemical shielding anisotropies, ²⁹Si MAS NMR spectra of 1 can report on the molecular symmetry of 1 as well as $-$ selectively $-$ on Si(SiMe₃)₃ reorientational processes, whereas 13 C MAS NMR spectra of 1 will be affected by both internal R_3 Si and Si(SiMe₃)₃ reorientation [6]. With this aspect in mind, we first consider the ²⁹Si MAS NMR spectra of 1 (Fig. 1). At $T = 150$ K we observe three barely resolved ²⁹Si resonances ($\delta_{\text{iso}} = -9.3, -9.9, -10.2$ ppm), corresponding to three nonequivalent SiMe₃ groups, one resonance for the SiPh₃ group (δ_{iso}) -11.7 ppm), and one resonance for the central Si atom ($\delta_{\rm iso} = -133.4$ ppm). ²⁹Si CP NMR spectra of 1, obtained under static conditions, display a 29 Si chemical shielding tensor with a large asymmetry parameter $(-\delta_{iso} = \sigma_{iso} = 133.4$ ppm, $\sigma_{11} = 124.0$ ppm, $\sigma_{22} = 132.6$ ppm, $\sigma_{33} = 143.4$ ppm, $\eta = 0.85$) for the resonance of the central Si atom (see Fig. 1). Accordingly, solid 1 lacks all molecular symmetry and, if present, we should be able to detect $Si(SiMe₃)₃$ reorientation from one- and two-dimensional variable temperature 29Si MAS NMR experiments. Figure 2a shows the SiMe₃/SiPh₃ region of the ²⁹Si MAS NMR spectra of 1 in the

Fig. 1. ²⁹Si CP NMR experiments on 1 ($\omega_0/2\pi = 59.6$ MHz): a) ²⁹Si CP MAS NMR ($\omega_{\text{rot}}/$ $2\pi = 4.0$ kHz, $T = 150$ K); b) ²⁹Si CP NMR, static conditions, only the chemical shielding powder pattern of the 29Si resonance of the central Si atom at three different temperatures is shown

Fig. 2. Variable temperature ²⁹Si CP MAS NMR spectra of 1; only the SiPh₃/SiMe₃ region is shown, temperatures are indicated $(\omega_0/2\pi = 59.6 \text{ MHz})$, $\omega_{\text{rot}}/2\pi = 4.0 \text{ kHz}$: a) upper traces represent experimental spectra, lower traces are the corresponding best-fit calculated spectra, assuming minor intrinsic temperature-dependence of $\delta_{iso}({}^{29}\text{Si})$ of the four ^{29}Si resonances of the SiPh₃ and SiMe₃ groups in 1; b) calculated hypothetical ²⁹Si MAS NMR spectra of 1, assuming mutual exchange of the three SiMe_3 groups by $\text{Si}(\text{SiMe}_3)_3$ reorientation

temperature range of $T = 150-290$ K. It is tempting to ascribe the minor changes of isotropic chemical shifts in the SiMe₃ region, observed as a function of temperature, to $Si(SiMe₃)$ ₃ reorientation around the $Ph₃Si-Si_{central}$ bond direction and leading to coalescence of the three ^{29}Si resonances of the three $SiMe₃$ groups at higher temperatures. This explanation can be ruled out by calculating expected exchange broadened ²⁹Si NMR spectral lineshapes for mutual exchange of the three SiMe₃ groups in 1 and comparing them to the experimental ²⁹Si MAS NMR spectra. Temperature dependent spectral lineshapes caused by $Si(SiMe₃)₃$ reorientation around the Ph₃Si-Si_{central} bond direction would have to display an asymmetric shape in the spectral SiMe_3 region with a shoulder at higher frequencies at temperatures approaching coalescence (Fig. 2b), whereas the experimentally observed ²⁹Si spectral lineshapes display an inversely asymmetric

shape in this temperature region with an unresolved shoulder towards lower frequencies. Bearing in mind that the entire $\sin M_e$ region in the ²⁹Si MAS NMR spectra of 1 ($\omega_0/2\pi$ = 59.6 MHz) only spans a frequency range of approximately 100 Hz and that we consider a temperature range of roughly 150 degrees, it is reasonable to ascribe the observed changes in 29 Si spectral lineshapes to minor intrinsic temperature shifts of $\delta_{iso}(^{29}\text{Si})$. With this assumption, the temperature dependent 29Si NMR spectral lineshapes can be well reproduced (Fig. 2a); the simulated spectral lineshapes correspond to intrinsic temperature shifts of $\delta_{iso}({}^{29}\text{Si})$ for the SiMe₃ and SiPh₃ groups in solid 1 of 0.003, 0.006, 0.005 ppm/K (SiMe₃) and 0.007 ppm/K (SiPh₃). Furthermore, within experimental error the ²⁹Si chemical shielding tensor of the resonance of the central Si atom does not change as a function of temperature (see Fig. 1), and also ²⁹Si 2D EXSY experiments on 1 (data not shown) under MAS ($T = 204$ and 224 K, mixing times of $\tau_{\text{mix}} = 4$ and 3 s) and under static conditions ($T = 290$ K, $\tau_{mix} = 5$ s and 50 µs) give no indication for the presence of a molecular dynamic process leading to mutual exchange of the three crystallographically nonequivalent SiMe_3 sites in 1. In short, on the frequency scales accessible by one- and two-dimensional variable temperature 29 Si MAS NMR experiments $(10^{-2}-10^{3} s^{-1})$ on solid 1, the molecule in the crystal lattice does not undergo $Si(SiMe₃)$ ₃ reorientational jumps around the Ph₃Si-Si_{central} bond direction. The absence of this kind of dynamic disorder in solid 1 is in contrast to the dynamic properties of the (chemically) symmetrically substituted derivative $Si(SiMe₃)₄$. For the latter, whole-molecule reorientational jumps do occur at temperatures of $T \approx 170-220$ K with exchange rate constants of the order of $k \approx 10^{-1} - 10^2$ s⁻¹ [6].

Having established the complete lack of molecular symmetry and the absence of Si(SiMe₃)₃ reorientation for solid 1 by means of ²⁹Si MAS NMR, we may next inspect the solid-state dynamic properties of 1 as seen by one- and twodimensional variable temperature 13 C MAS NMR. The lack of molecular symmetry for 1 leads us to expect up to eighteen resonances in the phenyl region and nine resonances in the methyl region of the 13 C MAS NMR spectrum of 1. Not so surprisingly, even at $T = 150$ K (Fig. 3) these multiple carbon sites are only

Fig. 3. ¹³C CP MAS NMR spectra of 1 ($\omega_0/2\pi = 75.5$ MHz, $\omega_{\text{rot}}/2\pi = 4.0$ kHz) at different temperatures; left column displays 13C-phenyl spectral region, right column 13C-methyl spectral region

partially resolved in the ¹³C MAS NMR spectrum ($\omega_0/2\pi = 75.5$ MHz). As far as the phenyl-¹³C region is concerned, hardly any changes are observed in the ¹³C MAS NMR spectra of 1 as a function of temperature. On the timescale of onedimensional ¹³C MAS NMR experiments in the temperature range of $T = 150-$ 290 K, we have no indication for the occurrence of either reorientation of the SiPh₃ group around the Ph₃Si-Si_{central} bond direction or for π -flips of the individual phenyl groups in the SiPh₃ moiety. A different situation is found for the methyl-¹³C region for which we do observe a strong temperature dependence in the 13 C MAS NMR spectra of 1 (Fig. 3). The only remaining possible dynamic process to account for these temperature dependent ¹³C MAS NMR effects is internal $2\pi/3$ reorientation of the three crystallographically inequivalent SiMe₃ moieties in solid 1. Given the incomplete spectral resolution in the methyl-¹³C region of these ¹³C MAS NMR spectra, it would be a rather hopeless task to immediately attempt quantitative analysis of these spectral lineshapes in terms of exchange rate constants of internal SiMe₃ reorientation as a function of temperature by means of spectral lineshape simulations. There are too many unknown parameters, such as the unknown assignment of the nine incompletely resolved 13 C resonances into groups of three, each group representing one SiMe_3 unit. Furthermore, with the three SiMe_3 groups in solid 1 being crystallographically inequivalent, these three SiMe₃ moieties do not necessarily have to undergo internal SiMe₃ reorientation with identical exchange rate constants $k, k',$ and k'' . Unequal exchange rate constants of internal $2\pi/3$ reorientation for crystallographically inequivalent [']Bu groups in a Si^tBu₃ moiety [10] or SiMe₃ groups in a $\overline{C(SiMe_3)_3}$ unit [11] have previously been observed.

Here we can take advantage of $2D¹³C$ MAS NMR EXSY experiments. As an example, the contour plot of a two-dimensional 13 C MAS NMR exchange experiment of 1 at $T = 169$ K and with a mixing time $\tau_{\text{mix}} = 250$ ms is displayed in Fig. 4. 2D ¹³C EXSY experiments on 1, obtained with different mixing times τ_{mix} and at various different temperatures, permit unambiguous assignment of the

Fig. 4. Contour plots of 2D ¹³C MAS EXSY spectra of 1 ($T = 169$ K, $\tau_{\text{mix}} = 250$ ms); only the ¹³Cmethyl region is shown; a) experimental spectrum, b) best-fit simulated spectrum

methyl- 13 C spectral region into three groups of three methyl- 13 C resonances, each representing one SiMe_3 group. We already know that the only molecular dynamic process we have to take into account for 1 is internal $2\pi/3$ SiMe₃ reorientation. Accordingly, off-diagonal peaks in the methyl-¹³C spectral region in 2D ¹³C EXSY spectra of 1 can only occur between methyl $1³C$ resonances belonging to methyl groups within the same SiMe₃ group. Careful inspection of 2D ¹³C EXSY spectra of 1 (Fig. 4) reveals these " 3×3 assignment connectivities" and thus provides valuable information input for subsequent spectral lineshape simulations of the one-dimensional variable temperature 13 C MAS NMR spectra of 1. Beyond this assignment aspect from $2D¹³C EXSY$ spectra of 1, differing relative off-diagonal intensities for ¹³C-methyl resonances belonging to different SiMe₃ groups directly and clearly indicate unequal exchange rate constants k, k', and k'' for internal $2\pi/3$ SiMe_3 reorientation of the three crystallographically inequivalent SiMe_3 groups in solid 1.

Most commonly, $2D¹³C EXSY$ experiments are carried out at temperatures low enough to be able to operate in the so-called slow-exchange regime. In this regime, exchange rate constants of the dynamic process under consideration are small enough to allow to take into account only exchange occuring during the mixing time of the two-dimensional NMR experiment and to neglect all motion occuring during the evolution and detection period. 2D ¹³C EXSY NMR data obtained in this slow-exchange regime yield exchange rate constants in a straightforward manner directly from the relative integrated intensities of off-diagonal peaks as a function of the mixing time τ_{mix} ; there is no need for a full simulation of the two-dimensional NMR experiment in order to extract kinetic parameters. For $13C$ NMR experiments on solid 1, this slow-exchange regime is only reached at inconveniently low temperatures (T < 160 K). Hence, 2D ¹³C EXSY experiments on 1 have to be obtained in the so-called intermediate-exchange regime where exchange occuring during the evolution and detection period can no longer be neglected for meaningful analysis. In order to extract exchange rate constants from 2D 13C EXSY experiments carried out in the intermediate-exchange regime, a full simulation of the two-dimensional NMR experiment, including the effects of exchange occuring during evolution and detection period, is necessary. Methods to fully analyze and simulate such experimental 2D EXSY data have been described in detail in the literature [12]; the contour plot of the best-fit simulated 2D 13 C EXSY spectrum of 1 is shown in Fig. 4b in comparison to the contour plot of the corresponding experimental spectrum. Qualitatively, it is obvious from inspection of the 2D ¹³C EXSY experiment that internal SiMe_3 reorientation for one of the three SiMe_3 groups in 1 occurs at a lower exchange rate than for the other two $SiMe₃$ groups in the molecule. Of course, full simulation of such intermediateregime $2D$ ¹³C EXSY experiments yields exchange rate constants.

Based on the 2D 13 C EXSY information (signal assignment as well as knowing that internal SiMe₃ reorientation of one of the three SiMe₃ groups occurs at a considerably slower rate than for the other two SiMe_3 groups), we can now proceed and simulate the one-dimensional variable temperature 13 C MAS NMR exchange broadened lineshapes in terms of exchange rate constants of internal SiMe_3 reorientation. Formally, the exchange broadened lineshapes can be described as a superposition of three (independent) three-site exchange processes. The results

Fig. 5. Variable temperature ¹³C MAS NMR spectra of 1; only the ¹³C-methyl region is shown and temperatures are indicated. Upper traces represent experimental spectra, lower traces are the corresponding best-fit calculated spectra; the bar code at the bottom indicates assignment of the ^{13}C methyl resonances into groups of three, each group representing one $\sin M$ e₃ group; the smaller linewidth of the bars for one SiMe_3 group indicates slower exchange than for the other two SiMe_3 groups (see also Fig. 6)

of this iterative fitting procedure are shown in Fig. 5 where the 13 C-methyl region of experimental 13 C MAS NMR spectra of 1 in the temperature range of $T = 150 - 225$ K is compared to the best-fit calculated spectral lineshapes. The semilogarithmic plot of exchange rate constants vs. T^{-1} in Fig. 6 summarizes the exchange rate constants for internal SiMe_3 reorientation in solid 1 as determined by simulation of one- and two-dimensional variable temperature 13 C MAS NMR experiments. Over the temperature range considered, the exchange rate constants follow Arrhenius' law, and from the slope of the plots depicted in Fig. 6 we obtain activation energies $E_a = 21 \pm 4$, 22 ± 4 , and 30 ± 5 kJ mol⁻¹, respectively, for the internal reorientation of the three $SiMe₃$ groups in 1. For $Si(SiMe₃)₄$, we find $E_a = 53 \pm 6$ kJ mol⁻¹ for internal SiMe₃ reorientation [6]. For Si(SiMe₃)₄, the crystal structure has been determined [7], but not for 1. Accordingly, any attempts

Fig. 6. Semilogarithmic plots of exchange rate constants $k, k',$ and k'' for internal SiMe₃ reorientation in 1 vs. 1/T; the symbols refer to exchange rate constants determined from ¹³C MAS (*, \diamondsuit , \circlearrowright) and ¹³C 2D EXSY (\times , ∇ , \square) NMR experiments; note the difference in exchange rate constants for the three inequivalent SiMe_3 groups in solid 1

to rationalize these differences in E_a for internal SiMe₃ reorientation necessarily have to be speculative. A major contribution towards energy barriers for internal SiMe₃ reorientation of predominantly intramolecular nature seems a reasonable assumption, and thus implies a lesser degree of intramolecular steric hindrance of the SiMe₃ groups in 1 as compared to $Si(SiMe₃)₄$.

Experimental

Compound 1, $(\text{Ph}_3\text{Si})\text{Si}(\text{SiMe}_3)$ ₃, was synthesized following a published procedure [13].

NMR experiments and data analysis

All 13C and 29Si NMR experiments were carried out on a Bruker MSL 300 NMR spectrometer at *Larmor* frequencies of 75.5 (¹³C) and 59.6 (²⁹Si) MHz, respectively. Standard 4 mm and 7 mm ZrO₂ rotors and double-bearing probes were used. For variable temperature MAS NMR experiments, purified N_2 gas served as drive and bearing gas. MAS frequencies were in the range of 2.0–4.5 kHz, ¹H π /2 pulse durations were 3-5 µs, recycle delays were 3-5 s, and contact times for *Hartmann-Hahn* cross-polarization were 1.5 ms (¹³C) and 2.0 ms (²⁹Si). 2D exchange spectroscopy (EXSY) [14] employed phase cycling according to the TPPI method [15]. Spectral lineshape simulations of the exchange broadened one-dimensional 13 C and 29 Si MAS NMR spectra of 1 were carried out as described elsewhere [6, 10, 14, 16]. Some ¹³C 2D EXSY spectra of 1 have been obtained in the socalled intermediate motional regime. Simulation of such 2D spectra hence requires a suitably extended exchange matrix formalism as described in the literature [12] where also motion during the evolution and detection periods of the experiment is taken into account and where T_1 relaxation during the mixing time is still negligible. Iterative fitting procedures employed the MATLAB [17] simplex and quasi-Newton routines for nonlinear least-squares minimization.

Isotropic chemical shifts δ_{iso} (¹³C) and δ_{iso} (²⁹Si) are given with respect to external SiMe₄, chemical shielding tensor eigenvalues are reported according to *Haeberlen's* convention [18]; $\delta_{\rm iso} = -\sigma_{\rm iso}; |\sigma_{33} - \sigma_{\rm iso}| \ge |\sigma_{11} - \sigma_{\rm iso}| \ge |\sigma_{22} - \sigma_{\rm iso}|; \eta = (\sigma_{22} - \sigma_{11}) (\sigma_{33} - \sigma_{\rm iso})^{-1}.$

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